## C402: Chemical Rate Processes

## **Kinetic Measurements**

General features of fast reactions; study of fast reactions by flow techniques, relaxation methods (T-Jump, P-Jump, ultrasonic, pulse radiolysis, NMR); flash photolysis; salt and solvent effects on reactions in solutions. (5)

**Chain Reactions**: Features of chain reactions; thermal and photochemical reactions (hydrogen-bromine reaction, decomposition of aldehydes and ketones). (5)

**Kinetics of oscillatory reactions**: Introduction to oscillatory reactions; Belousov-Zhabotinsky and Field-Koros-Noyes models. (4)

**Rate Theory:** Concept of potential energy surfaces, transition state theory including its statistical mechanical treatment, Phenomenological theories of unimolecular reactions (Lindemann, Hinshelwood), statistical mechanical theories of unimolecular reactions (RRKM). (10)

**Reaction Dynamics**: Collision theory and Reaction Dynamics, Reaction Cross section and rate constant, Brief idea of Molecular Beam Scattering, Dynamics in condensed phase. (10)

**Femtochemistry**: Concepts and perspectives; applications to studies of dynamics and control of chemical reactions. (6)

## Recommended Books

- 1. Physical Chemistry, I. Levine, Tata McGraw Hill, 5<sup>th</sup> Edn., 2007.
- 2. Physical Chemistry: A Molecular Approach, D. A. McQuarrie and J. D. Simon, University Science Books, 1997.
- 3. Chemical Kinetics and Dynamics, J. I. Steinfeld, J. S. Francisco and W. L. Hase, Prentice Hall, 1999.
- 4. Chemical Dynamics in Condensed Phases: Relaxation, Transfer and Reactions in Condensed Molecular Systems, A. Nitzan, Oxford Univ. Press, 2006.

## **Reference Books**

- 1. Basic Chemical Kinetics, H. Eyring, S. H. Lin and S. M. Lin, John Wiley & Sons, New York, 1980.
- 2. The World of Physical Chemistry, K. J. Laidler, Oxford University Press, 1993.